Infrared Longitudinal Bands in Crystalline Carbon Dioxide

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Recently two absorption peaks at $678.3 \, \mathrm{cm^{-1}}$ and $2383.0 \, \mathrm{cm^{-1}}$ were reported for the cubic crystal of carbon dioxide. They appeared at non-normal incidence and disappeared at normal incidence.¹⁾ An attempt^{1,2)} has been made to interpret them as the longitudinal modes of v_2 and v_3 vibrations by means of the Haas-Hornig equation.³⁾ Assignment of these bands, however, still seems uncertain, because of the uncertainty of the value $\mathrm{d}\mu/\mathrm{d}Q$ used in the Haas-Hornig equation.

We have remeasured the absorption spectra of polycrystalline CO_2 films deposited fairly rapidly (60 \sim 90 sec) at liquid nitrogen temperature, the thickness being 1.8—2.8 μ . Spectra of each film were run at normal incidence, 0°, and at angles up to 30° from

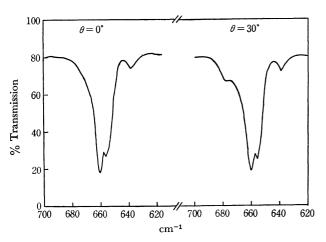


Fig. 1. Infrared absorption spectra of CO_2 crystal in the ν_2 region, at normal and non-normal incidence (unpolarized).

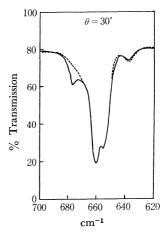


Fig. 2. Polarized absorption spectra of CO_2 crystal in the v_2 region at non-normal incidence;

—— p-polarized, ----- s-polarized.

normal. The results are completely in agreement with those reported by Parker and Eggers.1) The peaks at 678.0 cm⁻¹ and 2379 cm⁻¹ are angle-dependent, but the others do not shift or change significantly with the angle of incidence. We have further tried to confirm the longitudinal modes by a direct polarization measurement at non-normal incidence. As shown in Figs. 1 and 2, the peak at 678.0 cm^{-1} in the v_2 region observed at non-normal incidence appears strongly in the ppolarized component of the radiation, where the electric vector is parallel to the incident plane, disappearing in the s-polarized radiation where the electric vector is perpendicular to the incident plane. The absorption curve observed at normal incidence shows no difference in p- and s- polarizations. The same holds for the peak at 2379 cm⁻¹ in the v_3 region.

Since the peaks occur only in the p-polarization, this strongly supports their assignment to the longitudinal optical modes.

¹⁾ M.A. Parker and D. F. Eggers, J. Chem. Phys., 45, 4354 (1966).

²⁾ D. C. McKean, ibid., 52, 6451 (1970).

³⁾ C. Haas and D. F. Hornig, ibid., 26, 707 (1967).